Asymmetric Synthesis. 39.1 Synthesis of 2-(1-Aminoalkyl)piperidines via 2-Cyano-6-phenyl Oxazolopiperidine

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The asymmetric synthesis of a series of 2-(1-aminoalkyl) piperidines using (-)-2-cyano-6phenyloxazolopiperidine 1 is described. LiAlH₄ reduction of 1 followed by hydrogenolysis led to the diamine 3. The same strategy applied to C-2-methylated compound 7 afforded [(2.5)-2methylpiperidin-2-yl]methanamine (9). Addition of lithium derivatives to the cyano group of 1 resulted in the formation of an intermediate imino bicyclic system (11a-c) which could be diastereoselectively reduced to substituted diamino alcohols 13a-c. The addition of an excess of PhLi to 1 in the presence of LiBr furnished disubstituted amine 19, the precursor of diphenyl-[(2*S*)-piperidin-2-yl]methanamine (**22**).

Introduction

An ever growing interest in 1,2-diamine compounds exists since this system is an integral feature in many compounds of pharmaceutical value and in chiral ligands for asymmetric synthesis.² Among this family 2-(aminomethyl)pyrrolidine and -piperidine derivatives are of particular interest.^{3,4} These compounds are generally prepared in optically pure form from the corresponding amino acids⁵ or amino esters;^{3a} neither of these methods allows the diastereoselective preparation of derivatives substituted on the carbons bearing the amine functions. Furthermore, the preparation of (aminomethyl)piperidine derivatives requires not easily available enantiomerically pure pipecolic acid as a starting material, in contrast with proline which is available in its enantiomeric forms.⁶

In a continuation of our studies on 2-cyano-6-phenyloxazolopiperidine $\mathbf{1}^{7}$, we were interested in the reactivity of nucleophiles toward the cyano group. Our first results showed that the reaction of organolithium or cuprate derivatives with 1 gave an imine which was reduced to

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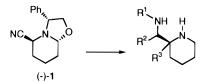


Figure 1.

a primary amine with a remarkable stereoselectivity.8 This reactivity has been used for the preparation of spiropiperidines related to histionicotoxin.9,10 We also observed that the aminonitrile function of alkylated derivatives of 1 could be reduced without elimination to give intermediate imines which can in turn be reduced to 1,2-diamines or react with electrophiles.⁹

We decided to investigate these reactivities in order to prepare unsubstituted and substituted aminomethyl piperidine derivatives (Figure 1).

Results and Discussion

Synthesis of (-)-[(2S)-piperidin-2-yl]methanamine (3). It was felt that synthon 1 was a good candidate for a short asymmetric synthesis of 2-(aminomethyl)piperidine, provided that the cyano group could be reduced cleanly without competitive elimination. It is known that α -aminonitriles can be reduced with LiAlH₄ either to diamines by reduction of the cyano group or into monoamines by a decyanation process. The results depend greatly on the substitution pattern.¹¹ Generally, monosubstituted α -aminonitriles yield diamines. Recently an example of reduction of 2-cyanopiperidine leading to a diamine has been published.¹²

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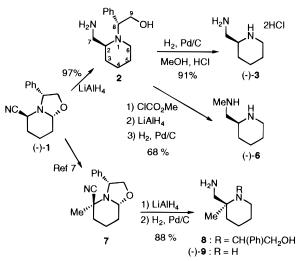
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2-Cyano-6-phenyloxazolopiperidine (1) was easily prepared from (*R*)-(–)-phenylglycinol and glutaraldehyde in the presence of KCN.⁷ When (–)-1 was treated with LiAlH₄ in Et₂O at rt, diamino alcohol **2** (Scheme 1) was obtained in 97% yield. As observed by Hoornaert,¹² the aminomethyl chain adopted a mixed axial–equatorial position ($\sum J_{2,3} = 11.5 \text{ Hz}$)¹³ ascribed to a strong gauche interaction between aminomethyl group and the *N*-(phenylethanol) substituent. Hydrogenolysis of the chiral appendage led easily to the 1,2-diamine **3** in 91% yield isolated as its dihydrochloride salt. Examination of the ¹H NMR spectrum indicated that an equatorial position was adopted by the aminomethyl chain ($\sum J_{2,3} > 14$ Hz).

To demonstrate the versatility of our strategy for preparing chiral ligands used in asymmetric synthesis, we decided to study the selective N-alkylation of the primary amine. For this purpose, diamino alcohol **2** was treated with methyl chloroformate followed by reduction of the resulting carbamate **4** with LiAlH₄ then hydrogenolysis of *N*-methyl intermediate **5**. Under these conditions *N*-methyl diamine **6** was obtained as an optically pure compound in 66% overall yield from synthon **1**.

Synthesis of [(2S)-2-methylpiperidin-2-yl]methanamine (9). In order to prepare C-2-substituted derivatives of (aminomethyl)piperidine, we first examined the reduction of C-2-methylated compound 7 obtained as a pure isomer by a deprotonation-alkylation sequence from synthon 1.7 At rt in Et₂O, compound 7 was quantitatively reduced to diamino alcohol 8. Reduction of the nitrile function was confirmed by the appearance of an AB spin system (δ 2.87) in the ¹H NMR spectrum assigned to H-7 protons. This result was unexpected as it is known that disubstituted aminonitriles are generally reduced to monoamines by elimination of the cyano group, especially when the cyano group and nitrogen lone pair are in an antiperiplanar disposition.¹⁴ In the 2-cyano-6-phenyloxazolopiperidine series, it is known¹⁵ that the cyano group adopts an axial position antiperiplanar to the nitrogen lone pair (Figure 2). It is likely that the

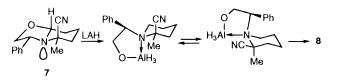
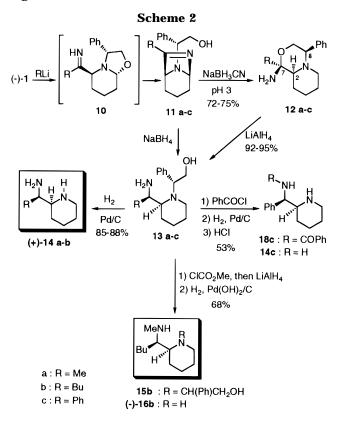


Figure 2.



oxazolidine function is reduced, leading first to a fivemembered chelating intermediate via the participation of the ring nitrogen. As a consequence, elimination of the nitrile cannot be assisted by the N-lone pair electron and an alternative mechanism; i.e reduction to the amino function prevails. Hydrogenolysis of **8** under standard conditions furnished C-2-methylated (aminomethyl)piperidine **9** in 90% yield.

Synthesis of (1-piperidin-2-yl)alkylamines 14a,b and 16b and (R)-phenyl[(2S)-piperidin-2-yl]methanamine (14c). The next target was the preparation of diamines substituted on the side chain (Scheme 2). A reinvestigation of the reaction of organolithium derivatives (MeLi, BuLi, or PhLi) with 1 indicated that the expected imines **10** were isolated in the cyclized form **11**¹⁶ as indicated by spectral analyses. All attempts to purify crude 11a led to decomposition products. A structural study was conducted on compound 11b. In MS (EI), the base peak was observed at m/z 255 (M - 31, M - CH₂-OH), indicating a product with a phenylethanol opened chain substituent on the nitrogen. In the ¹H NMR two narrow triplets were observed at δ 3.34 (J = 4.1 Hz) and 5.2 ppm (J = 2.5 Hz); they were attributed to bridgehead H-2 and H-6, respectively, by ¹H-¹³C COSY and are indicative of diaxial substituents at C-2 and C-6. The formation of these imines can be easily explained by the attack of an intermediate imine salt on the potential iminium ion at C-6.

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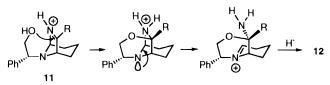
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Complete reduction of 11b with NaBH₄ in CH₃OH occurred with a remarkable stereoselectivity, giving the amino alcohol 13b. However, in the case of 11a and 11c the reaction was not diastereoselective. The problem was circumvented using a two-step route: (i) NaBH₃CN reduction of 11a-c at pH 3 affording 12a-c (ii) LiAlH₄ reduction of **12a-c** leading stereoselectively to compounds **13a**–**c**. Morpholines **12** were obtained as single isomers for R = Bu (12b) and R = Ph (12c) and as a 9/1 mixture of two isomers at C-7 for R = Me(12a). For 12b, 12c, and the major isomer of 12a, an axial position for H-2 and H-8 was indicated by similar chemical shifts and coupling constant values in their ¹H NMR spectra. A NOE experiment conducted on 12a (major) indicated a *cis* relationship for H-2, H-8, and the C-7 methyl group. Thus it was possible to assign the (2*S*,7*S*,8*R*) absolute configuration for compound 12a (major) and consequently the same for compounds 12b and 12c.

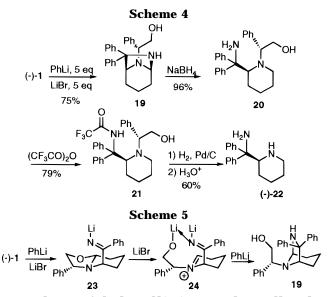
This configuration is explained by a preferential intramolecular attack of the primary alcohol of **11** on the accessible *si* face of the protonated imine, followed by the reduction of the aminal function of the intermediate, which could not be isolated (Scheme 3).

Stereoselective opening of the amino ether function of 12a-c with LiAlH₄ afforded exclusively 1,2-diamines 13a-c. The H-2, H-7 coupling constant values indicated an *erythro* stereochemistry.¹⁷ This configuration has been previously confirmed by the preparation of a rigid intermediate derived from diamine 14b.8 It is interesting to notice that the major products of the direct reduction of **11b** with NaBH₄ furnished compounds **13b** with the same configuration. This result indicates that reduction of morpholines 12 occurred via a retention mechanism as proposed for ketals¹⁸ and oxazolidines.¹⁹ Reduction with LiAlH₄ of the mixture of epimeric methylated derivatives 12a furnished exclusively erythro compound 13a. This observation could indicate a mechanism involving a transient iminium form with an intramolecular delivery of hydride. However, it might be possible that the major diastereomer of 12a is reduced much faster than the minor diastereomer.

Finally 1,2-diamines **14a** and **14b** were obtained by hydrogenolysis of the chiral appendage, in 59 and 50% overall yields, respectively, from **1**.

The synthesis of diamine **14c** containing a benzylamine function necessitated protection of **13c** as a benzamide **17c** before hydrogenolysis of the chiral phenylethanol moiety to give **18c**. Acid hydrolysis of the amide function of **18c** led finally to **14c**.

Compounds **13** represent useful intermediates allowing selective N-methylation *via* a two-step process involving an intermediate carbamate which was directly reduced (LiAlH₄) to compound **15b**. Hydrogenolysis of the chiral appendage afforded *N*-methyl diamine **16b**. Thus chemoselective substitution can be envisaged on each amino group.



Synthesis of diphenyl[(2S)-piperidin-2-yl]methanamine (22). When compound 1 was treated with an excess of PhLi, concomitant formation of 11c and diphenyl aminal 19 was observed (Scheme 4). Selective formation of 19 (75% yield) could be obtained using 5 equiv of PhLi and LiBr. This product was characterized by the absence of the imine absorption in the IR spectra. In the ¹³C NMR spectra a quaternary carbon was observed (δ 72.8 ppm), and in the ¹H NMR spectra, H-2 and H-6 appeared as broad singlets (δ 4.10 and 4.35). The mechanism of the reaction probably involves initial formation of imine salt 23 (Scheme 5), followed by association of LiBr, acting as a Lewis acid, allowing opening of the oxazolidine ring. Addition of PhLi on the intermediate iminium ion 24 gives the observed diphenyl aminal product 19. This hypothesis is supported by the non-reactivity of the unactivated monophenyl derivative 11c toward Ph-Li, even in the presence of LiBr. All attempts to introduce two different substituents during this reaction failed.

Generally the reaction of nitriles with organometallic species (RLi, RMgX), gives the corresponding imines.²⁰ Recently, Ciganek²¹ reported a double addition of alkylcerium dichlorides to nitriles to give tertiary carbinamines, while Wemple²² described the first example of double addition with the retention of chiral integrity of the asymmetric center α to the nitrile. However, to the best of our knowledge, this reaction has never been applied to α -aminonitrile in order to prepare 1,1-disubstituted 1,2-diamines.

NaBH₄ reduction of aminal **19** afforded diamino alcohol **20**. An examination of the ¹H NMR spectrum of **20** indicated that the piperidine ring of this compound adopts a boat conformation ($J_{2,3 \text{ ax}} = J_{2,3 \text{ eq}} = 7.5 \text{ Hz}$) which may be attributed to a strong gauche interaction between the C-2 substituent and the N-1 benzyl chain. Elimination of the chiral appendage of **20** required prior protection of the primary amine as a trifluoroacetamide **21**, followed by hydrogenolysis and then deprotection by acid treatment leading to **22** (34% yield from **1**).

Conclusion

We have achieved the enantiospecific synthesis of a series of piperidines bearing a 2-aminomethyl group which can be substituted at N and/or C atoms.

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These diamines constitute precursors for the elaboration of more sophisticated biologically interesting compounds. Furthermore the selective functionalization of the side chain allows the modulation of the catalytic properties of this series of chiral ligands.

Experimental Section

Infrared spectra were recorded as solution in CHCl₃. ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) were recorded in CDCl₃ or CD₃OD solution; chemical shifts were measured as ppm downfield of internal tetramethylsilane. Mass spectral data were recorded either in the electron-impact (EI) or chemical-ionization (CI) mode. Analytical TLC was performed on glass plates coated with silica gel 60 F₂₅₄ (Merck). Optical rotations of CHCl₃ or MeOH solutions were measured at 20 ± 3 °C. Tetrahydrofuran (THF) and ether were distilled from sodium/benzophenone immediately prior to use. All reactions were performed under an atmosphere of dry N₂.

(2R)-2-[(2S)-2-(Aminomethyl)piperidin-1-yl]-2-phenylethanol (2). To a stirred suspension of $LiAlH_4$ (0.5 g, 13.14 mmol) in Et₂O (25 mL) at -10 °C under N₂ atmosphere was slowly added a solution of 2-cyanopiperidine 1 (1 g, 4.38 mmol) in Et₂O (5 mL). After 2 h at rt the mixture was treated with 0.5 mL of H₂O, 0.5 mL of 15% aqueous NaOH, and 1.5 mL of H₂O. The white precipitate was filtered and washed several times with Et₂O. After removal of the solvent under reduced pressure, the compound 2 was isolated pure as a pale yellow oil in 97% yield (1.0 g, 4.25 mmol): $[\alpha]_{D} - 70$ (c 0.87, CHCl₃); MS (EI) m/z (rel intensity) 234 (M⁺, 1), 204 (80), 121 (10), 84 (80); ¹H NMR (CDCl₃) δ 1.0–1.65 (m, 6 H), 1.80 (ddd, J = 11.6, 11.0, 2.8 Hz), 2.46 (m), 2.82 (dd, J = 13.5, 2.8 Hz), 2.90 (dt, J= 11.6, 3.2 Hz), 3.30 (dd, J = 13.5, 5.0 Hz), 3.63 (dd, J = 10.5, 4.8 Hz), 4.02 (t, J=10.5 Hz), 4.27 (dd, J=10.5, 4.8 Hz), 7.12-7.35 (m, 5 H); ¹³C NMR & 23.1, 25.2, 28.8, 42.7, 44.7, 57.6, 60.4, 61.5, 126.7, 127.5, 128.2, 136.5; dihydrochloride (recryst MeOH/ Et₂O). Anal. Calcd for C₁₄H₂₄N₂OCl₂: C, 54.72; H, 7.87; N, 9.12. Found: C, 54.41; H, 7.64; N, 8.80.

[(2.5)-Piperidin-2-yl]methanamine (3). To a solution of amino alcohol **2** (0.9 g, 3.85 mmol) and 10% palladium on charcoal (150 mg) in MeOH (8 mL) was added 4 mL of MeOH–2 N HCl. The mixture was hydrogenated with stirring for 6 h. After filtration on Celite with MeOH, the solvent was removed under reduced pressure. The residue was washed several times with Et₂O to eliminate 2-phenylethanol. Crystallization of the dihydrochloride salt in MeOH/Et₂O gave **3** as white crystals in 91% yield (655 mg, 3.51 mmol): mp 240–242 °C; $[\alpha]_D - 5.7$ (*c* 0.42, MeOH); MS (CI) 115 (M + 1, 100), 84 (20); ¹H NMR (D₂O) δ 1.65, 1.95, 2.15 (3m, 6 H), 3.10 (td), 3.28 (dd, J = 13.6, 7.0 Hz), 3.40 (dd, J = 13.6, 5.8 Hz), 3.55 (m, 2H); ¹³C NMR (D₂O) δ 23.3, 23.8, 28.4, 43.5, 47.6, 56.2. Anal. Calcd for C₆H₁₆N₂Cl₂: C, 38.51; H, 8.61; N, 14.97. Found: C, 38.48; H, 8.71; N, 14.78.

(2R)-2-[(2S)-2-[(Methylamino)methyl]piperidin-1-yl]-2phenylethanol (5). A solution of amino alcohol 2 (1.15 g, 4.91 mmol) and methyl chloroformate (417 µL, 5.40 mmol) in CHCl₃ (15 mL) and 15% aqueous NaOH solution (3 mL) was stirred for 3 h at rt. The resulting mixture was washed with saturated aqueous NH₄Cl and extracted with CHCl₃. The combined organic layers were washed with water, dried over MgSO₄, and then evaporated to furnish the corresponding carbamate 4 as an oil which was dissolved in Et₂O (25 mL) and added to a stirred suspension of LiAlH₄ (0.38 g, 10 mmol) in anhydrous Et₂O (20 mL) at 0 °C. After stirring for 3 h at rt, the mixture was treated with H₂O (0.38 mL), 15% NaOH (0.38 mL), and then H_2O (1.14 mL). The resulting precipitate was filtered off and washed with Et₂O. After removal of the solvent under reduced pressure, methyl amine 5 was isolated as a colorless oil in 76% (0.914 g, 3.68 mmol): $[\alpha]_{\rm D}$ +15 (c 1.2, CHCl₃); IR 3415, 2936, 1686, 1543, 1109 cm⁻¹; MS (EI) m/z (rel intensity) 249 (MH+, 100), 218 (12), 204 (20), 141 (8), 121 (10), 105 (25), 91 (35), 84 (95). ¹H NMR (CDCl₃) δ 1.1–1.7 (m, 6H), 1.82 (ddd, J = 11.8, 10.5, 3.0 Hz), 2.47 (s), 2.62 (m), 2.71 (dd, J = 12.3, 3.5 Hz), 2.81 (dt, J = 11.8, 4.0 Hz), 3.00 (dd, J = 12.3, 4.8 Hz), 3.62 (dd, J = 10.3, 4.4 Hz), 3.95 (t, J =10.3 Hz), 4.22 (dd, J = 10.3, 4.4 Hz), 7.1–7.4 (m, 5H); ¹³C NMR (CDCl₃) δ 23.8, 25.9, 30.9, 37.0, 45.2, 54.3, 57.2, 61.0, 62.1,

127.4, 128.1, 128.7, 137.5; HRMS calcd for $C_{15}H_{25}N_2O~(MH^+)$ 249.1966, found 249.1963.

[(2.5)-Piperidin-2-yl]-*N***-methylmethanamine (6).** Hydrogenolysis of a solution of diamino alcohol **5** (0.85 g, 3.42 mmol) in MeOH (20 mL) in the presence of palladium on charcoal (0.15 g) afforded after filtration and evaporation of the solvent 0.84 g of an oily residue. 2-Phenylethanol was eliminated by trituration of the residue with Et₂O. Diamine **6** was obtained as an oil (0.394 g, 3.08 mmol) in 90% yield: $[\alpha]_D - 12$ (*c* 0.8, MeOH); IR 3425, 2933, 1647, 1450 cm⁻¹; MS (CI) 129 (M + 1), 98, 84; ¹H NMR (CDCl₃) δ 1.1–1.8 (m, 6H), 2.4 (s), 2.5–2.7 (m, 4 H), 3.10 (br d, J = 11.9 Hz); ¹³C NMR δ 24.6, 26.4, 30.7, 36.7, 46.7, 56.2, 57.8. Anal. Calcd for C₇N₂H₁₆: C, 65.57; H, 12.58; N, 21.84. Found: C, 65.40; H, 12.67, N, 21.74.

(2R)-2-[(2S)-2-(Aminomethyl)-2-methylpiperidin-1-yl]-2-phenylethanol (8). To a suspension of LiAlH₄ (1.0 g, 26.2 mmol) in ether (50 mL) was added a solution of compound 7^7 (1.0 g, 4.15 mmol) in ether. After stirring of the mixture for 3h, H₂O (0.5 mL) was carefully added, followed by NaOH (10%) (0.5 mL) and then water (1.5 mL). After filtration, concentration of the organic phases gave a solid which was crystallized from methanol (1.01 g, 98%): mp 252 °C; $[\alpha]_D$ +166 (c 1.0, MeOH); IR 3394, 2946, 1405, 1027 cm⁻¹; MS (EI) m/z (rel intensity) 249 (MH+, 10), 218 (50), 121 (25), 98 (100); ¹H NMR δ 0.70 (s), 1.05–1.80 (m, 6H), 2.48 (td, J = 11.2, 3.0 Hz), 2.72 (br s OH), 2.87 (AB system, 2 H), 2.95 (dt, J = 11.2, 3.6 Hz), 3.45 (dd, J = 10.6, 4.7 Hz), 3.85 (t, J = 10.6 Hz), 4.27 (dd, J = 10.6, 4.7 Hz), 7.1–7.4 (m, 5 H); 13 C NMR δ 18.7, 20.7, 27.1, 36.8, 40.8, 49.6, 57.2, 61.1, 62.1, 127.3, 129.3, 139.6. Anal. Calcd for C₁₅H₂₄N₂O: C, 72.54; H, 9.74; N, 11.28. Found C, 72.83, H, 9.86, N, 10.99.

[(2.5)-2-Methylpiperidin-2-yl]methanamine (9). Hydrogenolysis of 180 mg of **8** in the manner described in the preparation of **3** afforded **9** as hygroscopic crystals in 90% yield: $[\alpha]_D - 4.5$ (*c* 1.0, MeOH); IR 3421, 2923 cm⁻¹; MS (CI) 129 (MH⁺), 112, 102, 98; ¹H NMR (CD₃OD) δ 1.65 (s), 1.8–2.0 (m, 6H), 3.34 (br t, J = 5.9 Hz, 2H), 3.48 (br s, 2H); ¹³C NMR (CD₃OD) δ 18.6, 19.7, 22.6, 32.3, 41.2, 46.0, 57.1.

(2R)-2-[(1S,5S)-7-Methyl-6,8-diazabicyclo[3.2.1]oct-6en-8-yl]-2-phenylethanol (11b). A solution of 1 (2 g, 8.77 mmol) in freshly distilled Et_2O (30 mL) was cooled to -10 °C under a N2 atmosphere. After addition of n-BuLi (6.4 mL, 9.6 mmol), the resulting solution was stirred for 2 h at rt and then quenched with saturated aqueous NH₄Cl. The general workup procedure gave an orange oil which was purified by flash chromatography (SiO2, cyclohexane/AcOEt 1:1) to give nbutylimine **11b** (yellow oil, 2.1 g, 7.39 mmol, 85% yield): $[\alpha]_D$ -57 (c 1.5, CHCl₃); IR 3400, 1650, 1510 cm⁻¹; MS (EI) m/z(rel intensity) 286 (M⁺, 10), 255 (100), 203 (80); ¹H NMR $(CDCl_3) \delta 0.90$ (t, J = 7 Hz), 1.2–1.9, 2.0, 2.3 (m, 12 H), 3.34 (t, J = 4.1 Hz), 3.42 (dd, J = 5.5, 4.1 Hz), 3.75 (dd, J = 11.0, 4.1 Hz), 3.82 (dd, J = 11.0, 5.5 Hz), 5.2 (t, J = 2.5 Hz), 7.2-7.4 (m, 5 H); ¹³C NMR (CDCl₃) δ 13.8, 16.8, 22.7, 24.4, 25.3, 27.8, 30.6, 64.5, 66.1, 67.0, 85.6, 127.7, 128.4, 128.5, 140.5, 177.3

(2R)-2-Phenyl-2-[(1.5,5S)-7-phenyl-6,8-diazabicyclo[3.2.1]oct-6-en-8-yl]ethanol (11c). A solution of 1 (2 g, 8.77 mmol) in freshly distilled Et₂O (40 mL) was cooled to -78 °C. After addition of PhLi (5 mL, 10 mmol), the resulting solution was stirred for 1.5 h at rt and then guenched with saturated aqueous NH₄Cl. The general workup procedure gave an oily residue which was purified by flash chromatography (SiO₂, CH₂Cl₂/MeOH 95:5) to give imine **11c** as a yellow oil in 93% yield, (2.5 g, 8.15 mmol): $[\alpha]_D$ –43 (c 2.0, CHCl₃); IR 3350, 3027, 1640, 1520, 1428 cm⁻¹; MS (EI) *m/z* (rel intensity) 306 (M⁺, 14), 275 (100), 203 (45); ¹H NMR (CDCl₃) δ 1.1–1.9 (m, 6 H), 3.45 (dd, J = 5.3, 4.7 Hz), 3.81 (dd, J = 11.2, 4.7 Hz), 3.88 (dd, J = 11.2, 5.3 Hz), 3.95 (t, J = 2.9 Hz), 5.50 (t, J = 2.4 Hz), 7.3–7.6 (m, 10 H); 13 C NMR (CDCl₃) δ 16.8, 25.0, 25.2, 64.8, 65.7, 66.2, 86.2, 127.6, 127.9, 128.4, 128.5, 128.6, 131.0, 131.9, 140.4, 171.9; HRMS calcd for C₂₀H₂₂N₂O 306.1732, found 306.1725.

(1*S*,4*R*,9a*S*)-1-Methyl-4-phenyloctahydropyrido[2,1-*c*]-[1,4]oxazin-1-amine (12a). A solution of 1 (2 g, 8.77 mmol) in freshly distilled Et_2O (30 mL) was cooled to -10 °C. After addition of MeLi (6 mL, 9.60 mmol), the yellow solution was stirred for 2 h at rt and then guenched with saturated agueous NH₄Cl. The general workup procedure gave an orange oil which decomposed rapidly at room temperature. Direct solubilization with MeOH and THF (20 mL of each) gave a yellow solution which was acidified with 1 N aqueous HCl (pH 3). After addition of NaBH₃CN (580 mg, 9.23 mmol), the mixture was maintained at pH 3, refluxed for 1.5 h, and then neutralized at pH 7 with saturated aqueous NaHCO3 and extracted with CH₂Cl₂. The organic layer was dried on MgSO₄, concentrated, and purified by flash chromatography on silica gel (hexane/diethyl ether 1:1) to give a mixture of two epimeric compounds in 74% yield and in a 90/10 ratio. An analytical sample of **12a** was obtained by preparative TLC: $[\alpha]_D - 108$ $(c 0.46, CHCl_3)$; MS (EI) m/z (rel intensity) 246 (1), 245 (16), 231 (42), 212 (37), 187 (100), 186 (86), 104 (89); ¹H NMR (CDCl₃) & 1.35 (s, 3H), 1.4-1.8 (m, 7H), 2.1 (s, 2H, NH₂), 2.18 (dd, J = 10.8, 2.5 Hz), 2.72 (dt, J = 11.6, 3.6 Hz), 3.20 (dd, J= 11.0, 4.0 Hz), 3.48 (dd, J = 12.1, 4.0 Hz), 3.68 (dd, J = 12.1,11.0 Hz), 7.2-7.4 (m, 5 H); ¹³C NMR (CDCl₃) & 24.5, 25.5, 27.7, 26.4, 53.5, 65.3, 68.5, 69.1, 84.5, 127.5, 127.9, 128.5, 140.0. Anal. Calcd for C₁₅H₂₂N₂O: C, 73.13; H, 9.00; N, 11.37. Found: C, 72.97; H, 8.94, N, 11.18.

(1S,4R,9aS)-1-Butyl-4-phenyloctahydropyrido[2,1-c]-[1,4]oxazin-1-amine (12b). A solution of *n*-butylimine 11b (2 g, 6.99 mmol) in MeOH/THF (25 mL of each) was acidified with 1 N aq HCl (pH 3). After addition of NaBH₃CN (0.48 g, 7.69 mmol), the reaction mixture was maintained at pH 3, refluxed for 1.5 h, and then neutralized at pH 7 with saturated aq NaHCO₃ and extracted three times with CH₂Cl₂. The organic layers were dried over MgSO₄, concentrated, and purified by flash chromatography on silica gel (cyclohexane/ AcOEt 3:1) to give the butylmorpholine 12b in 72% yield (1.4 g, 4.86 mmol) as a pale yellow oil: $[\alpha]_D - 97$ (c 1.2, CHCl₃); IR 3400, 3020, 2938, 1480, 1224 cm⁻¹; MS (EI) m/z 288 (M⁺, 1), 272 (3), 231 (5), 187 (100); ¹H NMR (CDCl₃) δ 0.92 (t, J = 7Hz), 1.2-1.8 (m, 13 H), 2.10 (br s, NH₂), 2.20 (dd, J = 11, 2.1Hz), 2.70 (dt, J = 11.5, 2.5 Hz), 3.18 (dd, J = 11.1, 4.0 Hz), 3.46 (dd, J = 12, 4.0 Hz), 3.75 (dd, J = 12, 11.1 Hz), 7.2-7.4 (m, 5 H); ¹³C NMR (CDCl₃) δ 14.2, 23.3, 24.5, 24.7, 25.5, 27.3, 39.0, 53.6, 65.3, 67.2, 69.2, 85.6, 127.5, 127.9, 128.5, 140.2. Anal. Calcd for C18H28N2O: C, 74.95; H, 9.78; N, 9.71. Found: C, 75.01; H, 9.89; N, 9.76.

(1S,4R,9aS)-1,4-Diphenyloctahydropyrido[2,1-c][1,4]oxazin-1-amine (12c). Following the same procedure as for product 12a, phenylimine 11c (2 g, 6.53 mmol) was reduced with NaBH₃CN (0.45 g, 7.18 mmol) in 1.5 h at reflux under acidic conditions (MeOH/THF 1:1, 50 mL; 1 N HCl, pH 3). Identical workup and flash chromatography purification (SiO₂, hexane/ether 1:1) gave a white solid (12c) in 75% yield (1.5 g, 4.89 mmol): mp 155 °C (ether/heptane); $[\alpha]_D$ -50 (c 0.92, CHCl₃); IR 3400, 3090, 2950, 1510 cm⁻¹; MS (CI) 309 (M + 1, 100), 292 (35); ¹H NMR (CDCl₃) δ 1.0–1.8 (m, 7 H), 2.45 (dd, J = 11.0 and 2.4 Hz), 2.55 (br s, NH₂) 2.78 (br d, J = 11.2 Hz), 3.40 (dd, J = 11.2, 4.1 Hz), 3.68 (dd, J = 12.1, 4.1 Hz), 3.98 (dd, J = 12.1, 11.2 Hz), 7.3, 7.8 (m, 10 H); ¹³C NMR (CDCl₃) δ 24.6, 25.6, 26.9, 53.8, 65.6, 69.6, 69.8, 87.9, 127.3-128.7, 140.1, 143.6. Anal. Calcd for C₂₀H₂₄N₂O: C, 77.88; H, 7.84; N, 9.08. Found: C, 77.55; H, 7.78; N, 8.88.

(2R)-2-[(2S)-2-[(1R)-1-Aminoethyl]piperidin-1-yl]-2phenylethanol (13a). To a stirred suspension of LiAlH₄ (0.25 g, 6.57 mmol) in anhydrous Et₂O (25 mL) at -5 °C was added slowly a solution of methyl morpholine 12a as a mixture of isomers (1.4 g, 5.69 mmol) in ether (5 mL). After 1.5 h at rt, the mixture was treated with H_2O (0.25 mL), 15% aqueous NaOH (0.25 mL), and H₂O (0.75 mL). The resulting white precipitate was filtered and washed several times with Et₂O. After removal of the solvent under reduced pressure, compound 13a was isolated pure as a pale yellow oil in 95% yield (1.34 g, 5.4 mmol): $[\alpha]_D$ -68 (c 2.3, CHCl₃); IR 3010, 2890, 1600 cm⁻¹; MS (EI) m/z (rel intensity) 248 (M⁺, 2), 204 (100); (CI) 249, ¹H NMR (CDCl₃) δ 1.1 (d, J = 6.6 Hz), 1.2–1.6 (m, 6 H), 1.7 (td, J = 11.5, 2.5 Hz), 2.25 (dt, J = 10.8, 2.8 Hz), 2.6-2.7 (br s, NH₂, OH), 2.92 (br d, J = 11.5 Hz), 3.65 (dd, J =10.5, 4.8 Hz), 3.85 (qd, J = 6.6, 2.8 Hz), 4.08 (t, J = 10.5 Hz), 4.45 (dd, J = 10.5, 4.8 Hz), 7.1–7.4 (m, 5H); ¹³C NMR (CDCl₃) δ 20.5, 23.3, 24.0, 26.0, 44.6, 45.9, 59.7, 60.4, 61.9, 127.3, 128.0, 128.8, 136.1; HRMS calcd for $C_{15}H_{24}N_2O$ 249.1966, found 249.1964.

(2.*R*)-2-[(2.*S*)-2-[(1.*R*)-1-Aminopentyl]piperidin-1-yl]-2phenylethanol (13b). Aminoalcohol 13b (1.21 g, 4.15 mmol) was prepared from butylmorpholine 12b (1.3 g, 4.51 mmol) in 92% yield as described for 13a to give pale yellow crystals: mp 212–214 °C (Et₂O/hexane); $[\alpha]_D - 80$ (*c* 1.0, CHCl₃); IR 3400, 1510 cm⁻¹; MS (EI) *m*/*z* (rel intensity) 290 (M⁺, 20), 204 (100); ¹H NMR (CDCl₃) δ 0.95 (t, *J* = 7 Hz), 1.3–1.6 (m, 12 H), 1.8 (br t, *J* = 11.2 Hz), 2.35 (br d, *J* = 10.7 Hz), 2.75 (br s, NH₂, OH), 2.98 (br d, *J* = 11.2 Hz), 3.62 (m, 2 H), 4.10 (t, *J* = 10.4 Hz), 4.48 (dd, *J* = 10.4, 4.5 Hz), 7.1–7.4 (m, 5 H); ¹³C NMR (CDCl₃) δ 14.1, 22.9, 23.8, 24.2, 26.2, 29.2, 34.7, 46.2, 49.5, 59.8, 60.5, 60.9, 127.4, 128.0, 128.9, 136.4; HRMS calcd for C₁₈H₃₁N₂O 291.2434, found 291.2434.

(2*R*)-2-[(2*S*)-2-[(*R*)-(1-Aminophenyl)methyl]piperidin-1-yl]-2-phenylethanol (13c). Aminoalcohol 13c (1.27 g, 4.12 mmol) was prepared from phenylmorpholine 12c (1.35 g, 4.38 mmol) in 94% yield as described for 13a to yield an oil: $[\alpha]_D$ –40 (*c* 1.0, CHCl₃); IR 3015, 2895, 1550, 1520 cm⁻¹; MS (EI) *m*/*z* (rel intensity) 310 (M⁺, 15), 279 (100), 204 (100); ¹H NMR (CDCl₃) δ 0.8–1.9 (m, 7 H), 2.68 (dt, *J* = 10.7, 3.8 Hz), 3.0 (dt, 11.5, 2.5 Hz), 3.65 (dd, *J* = 10.8, 4.4 Hz), 3.9 (br s, NH₂, OH), 4.10 (dd, *J* = 10.8, 10.4), 4.5 (dd, *J* = 10.4, 4.4 Hz), 5.00 (d, *J* = 3.8 Hz), 7.1–7.5 (m, 10 H); ¹³C NMR (CDCl₃) δ 23.6, 23.9, 25.6, 45.7, 53.8, 60.8, 61.8, 62.4, 126.7, 127.2, 127.9, 128.4, 128.5, 129.0, 136.5; HRMS calcd for C₂₀H₂₆N₂O 310.2044, found 310.2042.

(1R)-1-[(2S)-Piperidin-2-yl]ethanamine (14a). A solution of amino alcohol 13a (0.95 g, 3.83 mmol) and 10% palladium on charcoal (0.15 g) in MeOH (10 mL) was acidified with 4 mL of MeOH/2 N HCl. The reaction mixture was stirred for 5 h under an H₂ atmosphere and then filtered over a Celite bed with MeOH and concentrated. After trituration with Et₂O, the resulting precipitate was washed several times $(Et_2O, 20 \text{ mL})$ to eliminate 2-phenylethanol. Recrystallization in MeOH/Et₂O led to white crystals of diamine 14a as a dihydrochloride salt (0.677 g, 3.37 mmol) in 88% yield: $[\alpha]_D$ +43 (c 0.56, MeOH); MS (CI) 201 (M + 1, 100), 112 (10), 84 (25); ¹H NMR (CD₃OD) δ 1.55 (d, J = 6.9 Hz), 1.7–2.2 (m, 6 H), 3.20 (td, J = 12.8, 3.8 Hz), 3.52 (ddd, J = 11.8, 5.1, 3.1), 3.60 (dt, J = 12.8, 1.6 Hz), 3.73 (qd, J = 6.9, 5.1 Hz); ¹³C NMR (CD₃OD) δ 15.7, 22.8, 25.6, 25.6, $\hat{4}6.8$, 50.6, 60.0. Anal. Calcd for C₇H₁₈N₂Cl₂: C, 41.79; H, 9.02; N, 13.92. Found: C, 42.04; H, 8.72; N, 13.85.

(1R)-1-[(2S)-Piperidin-2-yl]pentan-1-amine (14b). A solution of amino alcohol 13b (0.9 g, 3.10 mmol) and palladium hydroxide (20% on charcoal, 0.15 g) in MeOH (10 mL) was stirred for 6 h under an H₂ atmosphere and then filtered over Celite and concentrated under vacuo. The oily residue was purified by chromatography on alumina. The first elution with CH₂Cl₂/MeOH 98:2 gave the 2-phenylethanol byproduct, then CH₂Cl₂/MeOH/NH₄OH 85:10:5 led to butylamine 14b (0.44 g, 2.63 mmol) as a pale yellow oil (85% yield): $[\alpha]_D$ +6.3 (c 1.56, CHCl₃); IR 3288, 2930, 1590 cm⁻¹; MŠ (EI) m/z (rel intensity) 170 (M⁺, 100), 84 (35); ¹H NMR (CDCl₃) δ 0.90 (t), 1.2–1.6 (m, 12 H), 2.40 (ddd, J = 11.1, 4.0, 2.3 Hz), 2.62 (m), 2.65 (td, J = 11.9, 2.9 Hz), 3.10 (br d, J = 11.9 Hz); ¹³C NMR (CDCl₃) δ 14.1, 22.8, 24.8, 26.4, 26.7, 28.9, 33.4, 47.4, 55.7, 61.7. Anal. Calcd for C10H22N2: C, 70.52; H, 13.02; N, 16.45. Found: C, 70.37; H, 13.17; N, 16.35.

(2*R*)-2-[(2*S*)-2-[(1*R*)-1-(*N*-Methylamino)pentyl]piperidin-1-yl]-2-phenylethanol (15b). A solution of amino alcohol 13b (1.4 g, 4.82 mmol) in CHCl₃ (40 mL) was heated for 2 h under reflux with methyl chloroformate (0.41 mL, 5.31 mmol). The resulting mixture was quenched with saturated aqueous NH₄Cl and extracted following a general workup procedure to give a pale yellow oil which was dissolved in Et₂O (10 mL) and added to a stirred suspension of LiAlH₄ (0.16 g, 4.2 mmol) in anhydrous Et₂O (30 mL) at -5 °C. After 3 h at rt, the mixture was treated with H₂O (0.16 mL), 15% aqueous NaOH (0.16 mL), and H₂O (0.48 mL). The resulting white precipitate was filtered and washed several times with Et₂O. After removal of the solvent under reduced pressure, methyl amine 15b was isolated as a colorless oil in 78% yield (1.14 g, 3.76 mmol): [α]_D -105 (*c* 1.0, CHCl₃); IR 3250, 2930, 1450 cm⁻¹;

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MS (EI) m/z (rel intensity) 304 (M⁺, 12), 274 (10), 204 (100); ¹H NMR (CDCl₃) δ 0.95 (t), 1.35–1.7 (m, 13 H), 2.45 (dt, J = 11.1, 2.7 Hz), 2.50 (s), 2.97 (br d, J = 11.6 Hz), 3.14 (m), 3.62 (dd, J = 10.4, 4.9 Hz), 4.05 (dd, J = 10.9, 10.4 Hz), 4.28 (dd, J = 10.9, 4.9 Hz), 7.2, 7.35 (m, 5 H); ¹³C NMR (CDCl₃) δ 14.1, 23.2, 24.5, 24.8, 26.2, 28.6, 30.4, 35.0, 46.3, 58.6, 59.2, 59.7, 60.4, 127.5, 128.0, 128.9, 136.1; HRMS calcd for C₁₉H₃₂N₂O 304.2512, found 304.2515.

(1R)-N-Methyl-1-[(2S)-piperidin-2-yl]pentan-1-amine (16b). A solution of diamino alcohol 15b (0.5 g, 1.65 mmol) and 20% palladium hydroxide on charcoal (80 mg) in MeOH (10 mL) was stirred for 3 h under a hydrogen atmosphere and then filtered over a Celite pad and concentrated under vacuo. The oily residue was purified by flash chromatography (SiO₂). First elution gave the 2-phenylethanol byproduct (CH₂Cl₂/ MeOH, 95:5), then other solvent mixture (CH₂Cl₂/MeOH, 90: 10) led to the title compound 16b (0.26 g, 1.43 mmol) as a colorless oil in 87% yield: $[\alpha]_D - 12$ (c 1.0, CHCl₃); IR 3420, 2950, 2700, 1592, 1467 cm⁻¹; MS (EI) *m/z* (rel intensity) 185 (M + 1, 18), 151 (10), 137 (15), 123 (15); ¹H NMR (CDCl₃) δ 0.85 (t), 1.20–2.0 (m, 12 H), 2.60 (s), 2.90 (td, J = 12.4, 4.0 Hz), 3.12 (dt, J = 12.0, 2.0 Hz), 3.20 (m), 3.45 (brd, J = 12.4Hz); ¹³C NMR (CDCl₃) δ 13.9, 22.6, 22.6, 22.7, 22.9, 28.4, 28.4, 33.7, 45.4, 58.0, 61.3; HRMS calcd for $C_{11}H_{24}N_2$ 184.1938, found 184.1929

N-[(R)-[(2S)-1-[(1R)-2-Hydroxy-1-phenylethyl]piperidin-2-yl]phenylmethyl]benzamide (17c). To a solution of amino alcohol 13c (1.24 g, 4 mmol) in CH₂Cl₂ (60 mL) was added diluted aqueous NaOH (15%, 4 mL) and benzoyl chloride (0.62 g, 4.4 mmol). The reaction mixture was stirred for 2.5 h at rt and then diluted with saturated aqueous NH₄Cl and extracted with CH₂Cl₂. The organic layers were concentrated and purified by chromatography (Al₂O₃, CH₂Cl₂/MeOH 99:1) to give compound **17c** as a yellow oil (1.37 g, 3.32 mmol, 83% yield): $[\alpha]_D$ -38.4 (*c* 1.0, CHCl₃); IR 3332, 2931, 1637 cm⁻¹; MS (EI) m/z (rel intensity) 414 (M⁺, 15), 383 (100), 309 (25), 293 (15); ¹H NMR (CDCl₃) δ 0.9–1.8 (m, 6 H), 2.11 (td, J = 11.4, 2.8 Hz), 2.75 (ddd, J = 8.6, 5.2, 4.8 Hz), 3.08 (dt, J =11.4, 3.6 Hz), 3.52 (dd, J = 10.8, 5.3 Hz), 3.97 (dd, J = 10.8, 10.1 Hz), 4.40 (dd, J = 10.1, 5.3 Hz), 6.08 (dd, J = 7.1, 5.2 Hz), 6.82 (d, J = 7.1 Hz), 7.3–7.9 (m, 15 H); ¹³C NMR (CDCl₃) δ 23.5, 25.4, 25.6, 45.1, 52.7, 60.5, 61.7, 62.2, 126.2–131.6, 135.6, 135.9, 140.2, 167.7; HRMS calcd for C₂₆H₂₇N₂O (M CH₂OH) 383.2122, found 383.2116.

N-[(*R*)-Phenyl[(2.5)-piperidin-2-yl]methyl]benzamide (18c). A solution of amino alcohol 17c (1.2 g, 2.91 mmol) and activated 10% palladium on charcoal (180 mg) in MeOH (10 mL) was stirred for 3 h under an H₂ atmosphere and then filtered over Celite and concentrated. 2-Phenylethanol was eliminated by trituration of the oily residue with Et₂O. Pure compound 18c was thus obtained in 81% yield as a colorless viscous oil (0.69 g, 2.35 mmol): $[\alpha]_D$ +14 (*c* 1, CHCl₃); IR 3340, 2940, 1638 cm⁻¹; MS (EI) *m*/*z* (rel intensity) 295 (M + 1, 10), 174 (50), 122 (50), 105 (100); ¹H NMR (CDCl₃) δ 0.9–1.9 (m, 6 H), 2.51 (td, *J* = 13.0, 2.4 Hz), 2.84 (ddd, *J* = 10.7, 6.1, 1.8), 2.97 (br d, *J* = 13.0 Hz), 5.0 (dd, *J* = 7.8, 6.1 Hz), 7.1–7.9 (m, 10 H), 7.62 (d, *J* = 7.8 Hz); ¹³C NMR (CDCl₃) δ 24.6, 26.8, 30.1, 47.0, 57.8, 60.4, 126.5–131.3, 134.6, 139.2, 166.6; HRMS calcd for C₁₉H₂₂N₂O 294.1732, found 294.1748.

(*R*)-Phenyl[(2.5)-piperidin-2-yl]methanamine (14c). A solution of benzamide 18c (0.6 g, 2.04 mmol) in 6 N HCl (12 mL) was heated for 24 h under reflux and then diluted at rt and washed three times with Et₂O to eliminate benzoic acid. The aqueous layers were concentrated up to 5 mL, basified with NaOH (30%, pH 7), and extracted with CH₂Cl₂. The general workup procedure gave 14c as a pale yellow oil in 79% yield (0.3 g, 1.61 mmol): $[\alpha]_D$ +0.7 (*c* 1, MeOH); MS (EI) *m*/*z* (rel intensity) 191 (M + 1, 15), 172 (10), 153 (25), 106 (25), 84 (100); IR 3600, 2900, 1600, 1510 cm⁻¹; ¹H NMR (CDCl₃) δ 1.3–1.9 (m, 6 H), 2.62 (td, J = 12.1, 3.1 Hz), 2.80 (ddd, J = 10.9, 6.7, 2.2 Hz), 3.05 (dt, J = 12.1, 2.0 Hz), 3.9 (d, J = 6.7 Hz), 7.3 (m, 5 H); ¹³C NMR (CDCl₃) δ 24.4, 25.7, 28.0, 46.9, 60.4, 62.8, 127.2, 127.4, 128.5, 149.1; HRMS calcd for C₁₂H₁₉N₂ (M + 1) 191.1548; found 191.1559.

(2R)-2-[(1.5,5R)-7,7-Diphenyl-6,8-diazabicyclo[3.2.1]oct-8-yl]-2-phenylethanol (19). A solution of 1 (0.3 g, 1.31 mmol) and LiBr (0.592 g, 6.81 mmol) in freshly distilled Et₂O (20 mL) was stirred for 10 min at rt and then cooled to -20 °C before addition of phenyl lithium (3.4 mL, 6.8 mmol). The solution was slowly warmed to rt over 1.5 h and then quenched with saturated aq NH₄Cl. The combined organic layers were washed with H₂O, dried over MgSO₄, and concentrated. Purification by flash chromatography (SiO₂, CH₂Cl₂/MeOH 98: 2) led to compound **19** (0.380 g, 0.98 mmol) as a viscous oil in 75% yield: [α]_D -142 (*c* 1.7, MeOH); IR 3455, 3248, 2944, 2831, 1443, 1423 cm⁻¹; MS (EI) *m*/*z* (rel intensity) 384 (M⁺, 8), 367 (5), 353 (11), 307 (6), 289 (6), 265 (100); ¹H NMR (CDCl₃) δ 1.4–2.1 (m, 6 H), 3.80, 3.95 (2 m, 3 H), 4.10 (br s); 4.35 (br s), 7.2–7.6 (15 H); ¹³C NMR (CDCl₃) δ 15.6, 20.4, 25.6, 60.1, 62.0, 65.5, 71.8, 72.8, 125.7–128–4, 140.3, 140.9, 154.2. Anal. Calcd for C₂₆H₂₈N₂O: C, 80.21; H, 7.34; N, 7.29. Found: C, 80.21; H, 7.34; N, 7.03.

(2.R)-2-[(2.S)-2-(Aminodiphenylmethyl)piperidin-1-yl]-2-phenylethanol. (20). A methanolic solution (15 mL) of **19** (0.3 g, 0.78 mmol) and NaBH₄ (61 mg, 1.6 mmol) was stirred for 2.5 h at rt and then quenched with saturated aq NH₄Cl and extracted with CH₂Cl₂. The organic layer was washed with H₂O, dried over MgSO₄, and concentrated to yield 96% of pure amino alcohol **20** (0.291 g, 0.75 mmol) as an oil: $[\alpha]_D$ -101 (*c* 1.4, MeOH); IR 3521, 3196, 2366, 2333, 1428 cm⁻¹; MS (EI) *m*/*z* (rel intensity) 386 (M⁺, 17), 355 (3), 265 (17), 204 (100); ¹H NMR (CDCl₃) δ 0.65–1.15 (4 H), 1.6 (m), 1.80 (m), 2.0 (ddd, *J* = 14.4, 9.6, 7.2 Hz), 2.53 (ddd, *J* = 14.4, 4.3, 3.3 Hz), 3.72 (dd, *J* = 15.6, 8.2 Hz), 4.0 (m, 2 H), 4.22 (t, *J* = 7.5 Hz), 7.2–7.5 (m, 15 H); ¹³C NMR (CDCl₃) δ 19.7, 21.0, 23.7, 40.7, 63.5, 63.7, 66.8, 71.1, 126.0–129.0, 141.4, 144.1, 147.7; HRMS calcd for C₂₆H₃₀N₂O 386.2356, found: 386.2355.

N-[[(2S)-1-[(1R)-2-Hydroxy-1-phenylethyl]piperidin-2yl]diphenylmethyl]-2,2,2-trifluoroacetamide (21). A solution of amino alcohol 20 (1 g, 2.59 mmol) in CH₂Cl₂ (30 mL) was stirred for 4 h at 20 °C with trifluoroacetic anhydride (0.43 mL, 3.10 mmol) and triethylamine (0.73 mL, 5.18 mmol) and then quenched with saturated aq NH₄Cl. The general workup procedure gave an oily residue which was purified by flash chromatography (SiO₂, CH₂Cl₂). Pure compound 21 was isolated in 79% yield as a colorless oil (0.99 g, 2.04 mmol): $[\alpha]_D$ -18.4 (c 1, CHCl₃); IR 3420, 3250, 3100, 1750 cm⁻¹; MS (EI) m/z (rel intensity) 483 (M + 1, 25), 345 (20), 297 (30), 204 (100); ¹H NMR (CDCl₃) δ 1.2–1.7 (m, 7 H), 2.11 (br d, J =14.7 Hz), 3.87, 4.21 (2 br d, J = 12.0 Hz), 4.10 (dd, J = 4.0, 3.8 Hz), 4.42 (dd, J = 6.7, 6.0 Hz), 7.1–7.5, 7.6 (m, 15 H); ¹³C NMR (CDCl₃) & 18.3, 19.1, 21.3, 42.1, 60.8, 64.2, 67.6, 68.0, 127.4-129.8, 141.0, 141.2.

Diphenyl[(2S)-piperidin-2-yl]methanamine (22). A solution of trifluoroacetamide 21 (0.85 g, 1.76 mmol) in MeOH (12 mL) was acidified with concentrated HCl (4 mL) and stirred for 1 h under hydrogene atmosphere with 10% activated palladium on charcoal (100 mg). After filtration over Celite, the mixture was heated directly under reflux in diluted aqueous HCl (6 N, 20 mL) for 12 h. 2-Phenylethanol and trifluoroacetic acid byproducts were extracted of the crude mixture with CH_2Cl_2 . Aqueous layer was concentrated under vacuo (up to 5 mL), then neutralized (pH 7) and extracted following the usual work-up procedure. Pure diphenylamine 22 was then isolated as a colorless oil in 60% yield (0.28 g, 1.05 mmol): [a]_D -28.6 (c 0.5, CHCl₃); IR 3317, 3057, 2932, 2852, 1596 cm⁻¹; MS (EI) m/z (rel intensity) 267 (M + 1, 20), 250 (25), 182 (26), 165 (12), 104 (54), 85 (67), 84 (100); ¹H NMR (CDCl₃) δ 1.20–1.80 (m, 6 H), 1.95 (br s, NH, NH₂), 2.68 (td, J = 11.5, 2.7 Hz), 3.05 (br d, J = 11.5 Hz), 3.49 (dd, J = 10.5, 2.0 Hz), 7.35, 7.5 (m, 10 H); ¹³C NMR (CDCl₃) δ 25.4, 26.8, 27.1, 47.8, 63.5, 63.8, 126.8-129.2, 146.0, 146.8; HRMS calcd for C₁₈H₂₃N₂ 267.1860, found 267.1854.

Supporting Information Available: ¹H and ¹³C NMR spectra for compounds **5**, **9**, **11c**, **13a–c**, **14c**, **15b**, **16b**, **18c**, and **20–22** (26 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead for ordering information. JO960910S